





# New Palladacycles Derived Acylhydrazones and Hydrazones as Pre-catalyst in Mirozoki-Heck Coupling and Oxyarylations

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Keywords: palladacicle, arylation, oxyarylation

## INTRODUCTION

Palladacycle pre-catalyst are considered to be a source of palladium nanoparticles formed *in situ* and have been used with success, with low catalytic loading, in several coupling reactions.<sup>1,2</sup>

Herein we report the preparation of palladacycles **1** and **2** through electrophilic C-H activation of acylhydrazones and hydrazones by reacting them with  $\text{Li}_2\text{PdCl}_4$  in methanol.<sup>2</sup> Although some hydrazone-based palladacycle are known in the literature.<sup>2</sup> we describe for the first time their use as catalyst in Heck-Mizoroki arylation and oxyarylation reactions.

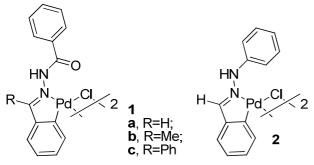
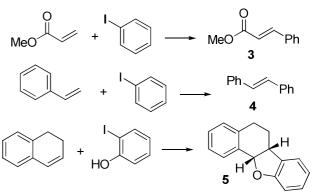


Figure 1. New palladacycles from hydrazones and acylhydrazones.

#### **RESULTS AND DISCUSSION**

Some of our selected results are described in Scheme 1, Table 1. The four palladacycle precatalyst showed in Figure 1 were used in low catalytic loading and promote the Heck-Mizoroki arylation of methyl acrylate by iodobenzene leading to **3** in good chemical yield (entries 1-4). Palladacycles **1b** and **1c** also promoted the arylation of styrene by iodobenzene in moderate chemical yields (entries 5 and 6). Finally, palladacycles **1a-c** catalyzed the oxyarylation of dihydronaphthalen by *ortho*-iodophenol in MeCN/H<sub>2</sub>O (1:3) in reasonable chemical yields (entries 7-9). Compound **5** could also be obtained in only 40 min, under microwave irradiation, although in this case the chemical yield decreased (entry 10).<sup>3</sup>



**Scheme 1.** Palladacycles **1** and **2** as precatalysts in Heck-Mizoroki arylation and oxyarylation reactions.

**Table 1.** Yields and main conditions for reaction showedin scheme 1.

| entry | catalyst        | conditions                        | products                    |
|-------|-----------------|-----------------------------------|-----------------------------|
| 1     | 1a <sup>°</sup> | DMF, TEA, 110 <sup>0</sup> C, 24h | <b>3</b> (80%)              |
| 2     | 1b*             | DMF, TEA, 110 <sup>0</sup> C, 24h | <b>3</b> (95%)              |
| 3     | 1c*             | DMF, TEA, 110 <sup>0</sup> C, 24h | <b>3</b> (95%)              |
| 4     | 2**             | DMF, TEA, 110 <sup>0</sup> C, 24h | <b>3</b> (40%) <sup>#</sup> |
| 5     | 1b*             | DMF, TEA, 110 <sup>0</sup> C, 24h | 4 (50%)                     |
| 6     | 1c*             | DMF, TEA, 110 <sup>0</sup> C, 24h | <b>4</b> (45%)              |
| 7     | 1a**            | MeCN/H <sub>2</sub> O, DIPEA,     | <b>5</b> (50%)              |
|       |                 | 120 <sup>0</sup> C, 24h           |                             |
| 8     | 1b**            | MeCN/H <sub>2</sub> O, DIPEA,     | <b>5</b> (55%)              |
|       |                 | 120 <sup>0</sup> C, 24h           |                             |
| 9     | 1c**            | MeCN/H <sub>2</sub> O, DIPEA,     | <b>5</b> (56%)              |
|       |                 | 120 <sup>0</sup> C, 24h           |                             |
| 10    | 1c**            | MeCN/H <sub>2</sub> O, DIPEA,     | <b>5</b> (37%)              |
|       | aalo/ **1 ma    | MW, 120 <sup>°</sup> C, 40 min.   |                             |

\*0.001 mol%. \*\*1 mol%. \*\* non-optimized result.

#### CONCLUSION

The new palladacycles described herein are promising and other synthetic applications are due in course in our laboratory.

### ACKNOWLEDGEMENTS

FAPERJ, CNPq, CAPES.

# REFERENCES

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14th Brazilian Meeting on Organic Synthesis – 14th BMOS – September 01-05, 2011-Brasilia, Brazil