

Search for an enzymatic approach to achieve the enantiomeric enrichment of β -borylated carboxylic esters

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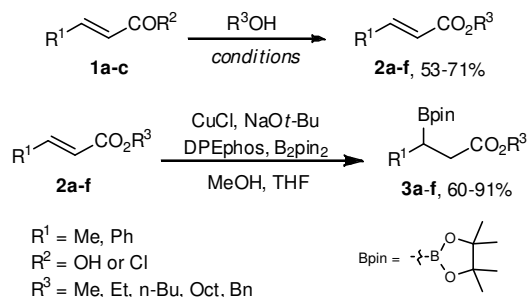
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INTRODUCTION

Lipases are the most used enzymes in organic chemistry. Their ability of enantiomers recognition by these biocatalysts represents a useful tool to synthesis of enantiomerically enriched compounds.¹ Additionally, boron-containing compounds have increased the interest of scientists for being important building blocks of bioactive substances.² The organoboron compounds are employed mainly in the Suzuki-Miyaura cross-coupling reactions.³ Herein, we are focused on development of an enzymatic way to enrich enantiomerically boron-containing carboxylic esters, catalyzed by lipases.

RESULTS AND DISCUSSION

Initially, the carboxylic esters were synthesized from acids or acyl chlorides. β -borylated compounds were synthesized according to the methodology described in the literature.⁴ (Scheme 1).



Scheme 1 – Syntheses of β -borylated compounds 3a-f.

Enzymatic assays were carried out in order to search of lipase for enantioselective hydrolysis of the compound 3a. Among tested catalysts, CAL-B has shown the highest activity (e.e. 78% for 3a); PSD-I and PSC-II furnished the ester with low e.e. (<10%). Although the product of hydrolysis is the carboxylic acid 4a-b, the efficiency of enzymatic assays was measured by e.e. values of compound 2a. We have also tested to perform the hydrolysis in organic solvents. Water-miscible and water-immiscible solvents were employed; however, no improvement was detected, leading low e.e. values (10-43%) for 3a. The highest e.e. was obtained when a mixture water:dioxane (9:1) was used as solvent. Our protocol was extended to compounds 3b-f, which present longer chains / steric hindrance.

Table 1 – Enzymatic hydrolysis of 3a-f.^a

$\begin{array}{ccc} \text{Bpin} & & \text{Bpin} \\ & & \\ \text{R}^1-\text{CH}=\text{CH}-\text{CO}_2\text{R}^3 & \xrightarrow[\text{700 rpm, 24 h}]{\text{CAL-B, H}_2\text{O, 32 }^\circ\text{C}} & \text{R}^1-\text{CH}(\text{Bpin})-\text{CH}_2-\text{CO}_2\text{R}^3 + \text{R}^1-\text{CH}(\text{Bpin})-\text{CH}_2-\text{CO}_2\text{H} \\ \text{3a-f} & & \text{3a-f} \quad \quad \quad \text{4a-b} \end{array}$			
Ester	R ¹	R ³	e.e. (%)
3a	Me	Me	78
3b	Me	Et	74
3c	Me	n-Bu	<10
3d	Me	Oct	rac.
3e	Me	Bn	rac.
3f	Ph	Et	rac.

^a Conditions: H₂O (1 mL), CAL-B (10 mg), 3a-f (0.05 mmol).

We have observed that these structural changes result in a large decrease in e.e. values. (Table 1). In view of achieve the enantiomeric enrichment of the boron compounds by lipase-catalyzed reaction, we turned our attention to transesterification reactions. Thereby, the reactions were carried out with esters 3a-c, using a set alcohol as solvent and CAL-B as catalyst. Unfortunately, the preliminary tests gave no product from transesterification reaction. (Table 2).

Table 2 – Transesterification reaction of 3a-c.^a

$\begin{array}{ccc} \text{Bpin} & & \text{Bpin} \\ & & \\ \text{R}^3-\text{CH}=\text{CH}-\text{CO}_2\text{R}^3 & \xrightarrow[\text{700 rpm, 24 h}]{\text{CAL-B, R}^4\text{OH, 32 }^\circ\text{C}} & \text{R}^3-\text{CH}(\text{Bpin})-\text{CH}_2-\text{CO}_2\text{R}^3 + \text{R}^3-\text{CH}(\text{Bpin})-\text{CH}_2-\text{CO}_2\text{R}^4 \\ \text{3a-c} & & \end{array}$			
Ester	R ³	R ⁴	
3a	Me	Et	
3b	Et	Me	
3c	n-Bu	Me	

^a Conditions: R⁴OH (1 mL), CAL-B (10 mg), 3a-c (0.05 mmol).

CONCLUSION

In summary, we have shown an enzymatic way to achieve the enantiomeric enrichment of β -borylated carboxylic esters. Esters presenting a long carbon chain did not suffer any hydrolysis reaction. The improvement of transesterifications are in progress.

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