

A Chemoenzymatic Approach to C1-C6 Fragment of (–)-Putaminoxin

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INTRODUCTION

(–)-Putaminoxin (**1**) is a decalactone isolated from the culture filtrates of the fungus *Phoma putaminum*, a leaf necrosis agent of the weed *Erigeron annuus*.^{1,2} These and other structure related compounds showed significant phytotoxic activity against several plants.^{1,3} Herein we describe our efforts to the synthesis of the C1-C6 fragment of the title compound [(*S*)-5-hydroxyhept-6-enoic acid (**2**)], using a kinetic enzymatic resolution approach as key step.

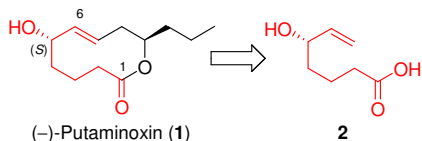
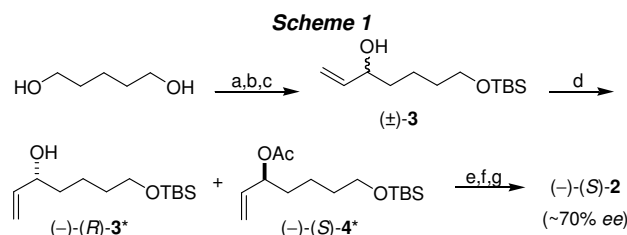


Figure 1. (–)-Putaminoxin (**1**).

RESULTS AND DISCUSSION

As previously described,⁴ the racemic allylic alcohol (\pm)-**3** could be prepared in 3 steps from commercial available 1,5-pentanediol (Scheme 1).



Reagents and conditions: (a) *i*: NaH, THF, r.t., 45 min; *ii*: TBSCl, r.t., 45 min (70%); (b) 2.5 eq. IBX, *tert*-butanol, 80 °C, 1 h (72%); (c) *i*: CH₂=CHMgBr, THF, 0 °C, 80 min; *ii*: NH₄Cl aq. (72%); (d) CALB, vinyl acetate, hexanes, 32 °C, 15 h, (–)-(*R*)-**3** (ee = 98%) and (–)-(*S*)-**4** (ee = 94%) (best condition; see Table 1); (e) chromatographic separation of (–)-(*S*)-**4**; (f) TBAF, THF, r.t., 4 h; (g) 0.4 eq. 2-iodobenzoic acid, 2.6 eq Oxone®, MeCN/H₂O (2:1), r.t., 6 h (yield not optimized). *The absolute configuration of the products were giving based on the enantioselectivity of the enzyme, but it is under investigation.

The kinetic enzymatic resolution of (\pm)-**3** was performed using the lipases CALB, *P. cepacia* [immobilized on ceramic (Amano PS-CII) and diatomite (Amano PS-DI)] or *P. fluorescens* (Amano AK) as biocatalysts (Table 1). The time of the KR is a critical point to obtain the products with good ee,

probably because the ratio of the reaction is similar to the two enantiomers of **3**.

Table 1. Kinetic Resolution (KR) of (\pm)-**3**.

Chemical reaction scheme showing the kinetic resolution of (\pm)-**3** to (–)-(*R*)-**3** and (–)-(*S*)-**4** using an enzyme in hexane at 32 °C.

Entry	Lipase	Time	(–)-(<i>R</i>)- 3 (% ee)*	(–)-(<i>S</i>)- 4 (% ee)*
1 ^a	CALB	24 h	99	91
2		15 h	98	94
3	<i>P. cepacia</i>	20 h	97	65
4	Amano PS CII	15 h	97	78
5	<i>P. cepacia</i>	20 h	>99	71
6	Amano PS-DI	15 h	99	79
7	<i>P. cepacia</i> Amano AK	20 h	22	27

*ee determined by chiral GC: β -cyclodextrin column, oven: 100–180 °C, rate: 2 °C/min.

From the results showed in Table 1 is noteworthy that lower reactional times led to high ee of (–)-**4**. Reaction of (\pm)-**3** with CALB using dimethylcarbonate (DMC), DMC/hexanes and MeCN as solvents led, in all cases, to recovery of starting material. The acetate (–)-**4** was deprotected by TBAF and the alcohol thus obtained was oxidized by catalytic IBX affording the acid (–)-**2** in 70% ee.

CONCLUSION

The C1-C6 fragment of the (–)-**1** could be prepared from commercial available starting materials using a chemoenzymatic approach. The use of new lipases and DKR (dynamic kinetic resolution) are under investigation by us, as well as the ring closing metathesis key step to the construction of the ten-membered ring of (–)-**1**.

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FAPESP, CNPq, CAPES

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