





Synthesis of new isoxazole derivatives with larvicidal activity

Diana C. B. da Silva-Alves, Janaína V. dos Anjos, Nery N. M. Cavalcante, Daniela M. do A. F. Navarro, Geanne K. N. Santos and Rajendra M. Srivastava*

Departamento de Química Fundamental, Universidade Federal de Pernambuco, Cidade Universitária, 50740-540, Recife, PE, Brazil

*Corresponding author: rms@ufpe.br

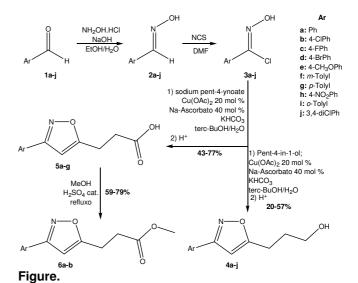
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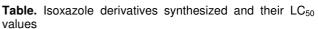
INTRODUCTION

One of the major routes of detoxification (removal of free radicals) in Aedes aegypti mosquitos is the kynurenine pathway. In this biochemical pathway, the most important reaction is the conversion of 3-hydroxykynurenine (3-HK), a harmful metabolite that leads to insect death, into xanthurenic acid, a nontoxic substance. This conversion is performed by the enzyme 3-hydroxy-kynurenine transaminase (HKT).¹ As a part of drug discovery program that has the main goal of finding new larvicide candidates, we have found that some heterocycles containing both aryl and propionic acid moieties may act as ligands for HKT, having larvicidal activity against A. aegypti species, the dengue vector. In this work, we describe the synthesis and the larvicidal evaluation tests for new isoxazole derivatives thar might becapable of accumulate 3-HK in insects, by inhibiting the enzyme HKT detoxification reaction.

RESULTS AND DISCUSSION

To synthesize the desired compounds via coppercatalyzed cycloaddition, we started with the imidoyl chlorides (3a-j) preparation. These compounds are transformed into the real dipoles, the nitrile oxides, in situ, by means of an inorganic base such as potassium bicarbonate. The imidoyl chlorides are obtained from corresponding aryloximes the (**2a-j**) and Nchlorosuccinimide (NCS), using DMF as solvent. For the synthesis of 3-(3-aryl-isoxazol-5-yl)-propan-1-ols (4a-j), 3a-j reacted with 4-pentyn-1-ol, using Cu(OAc)₂ and sodium ascorbate as catalytic system in the presence of potassium bicarbonate.² In a similar way, 3-(3-aryl-isoxazol-5-yl)-propionic acids (5a-g) were prepared in the reaction between 3a-j and sodium 4pentynoate using the same reaction conditions. Finally, we submitted two acids (5a-b) to classical esterification conditions in methanol, obtaining 6a-b as products (Figure). This modification was performed to verify what moiety is most important to maintain larvicidal activity: acid, alcohol or methyl ester. The larvicidal tests were performed according to procedures already reported in the literature.³ For these tests, we used A. aegypti larvae in the L4 growth stage. The results are summarized in the table and are expressed by means of LC₅₀ (lethal dose; dose which causes the death of 50% of the group test).





| 4a 54% >100 5a 63% 82.8 4b 20% 28.9 5b 50% 37.7 4c 33% >100 5c 77% 33.7 4d 57% 33.3 5d 45% 13.8 4e 20% >100 5e 43% 35.7 4f 57% >100 5f 52% >100 4g 52% >100 5g 48% 38.9 4h 44% 82.0 6a 59% 13.2 4i 22% >100 6b 79% 3.4 | Comp. | Yield | LC ₅₀ (ppm) | Comp. | Yield | LC ₅₀ (ppm) |
|--|-------|-------|------------------------|-------|-------|------------------------|
| 4c 33% >100 5c 77% 33.7 4d 57% 33.3 5d 45% 13.8 4e 20% >100 5e 43% 35.7 4f 57% >100 5f 52% >100 4g 52% >100 5g 48% 38.9 4h 44% 82.0 6a 59% 13.2 4i 22% >100 6b 79% 3.4 | 4a | 54% | >100 | 5a | 63% | 82.8 |
| 4d 57% 33.3 5d 45% 13.8 4e 20% >100 5e 43% 35.7 4f 57% >100 5f 52% >100 4g 52% >100 5g 48% 38.9 4h 44% 82.0 6a 59% 13.2 4i 22% >100 6b 79% 3.4 | 4b | 20% | 28.9 | 5b | 50% | 37.7 |
| 4e 20% >100 5e 43% 35.7 4f 57% >100 5f 52% >100 4g 52% >100 5g 48% 38.9 4h 44% 82.0 6a 59% 13.2 4i 22% >100 6b 79% 3.4 | 4c | 33% | >100 | 5c | 77% | 33.7 |
| 4f 57% >100 5f 52% >100 4g 52% >100 5g 48% 38.9 4h 44% 82.0 6a 59% 13.2 4i 22% >100 6b 79% 3.4 | 4d | 57% | 33.3 | 5d | 45% | 13.8 |
| 4g 52% >100 5g 48% 38.9 4h 44% 82.0 6a 59% 13.2 4i 22% >100 6b 79% 3.4 | 4e | 20% | >100 | 5e | 43% | 35.7 |
| 4h 44% 82.0 6a 59% 13.2 4i 22% >100 6b 79% 3.4 | 4f | 57% | >100 | 5f | 52% | >100 |
| 4h 44% 82.0 6a 59% 13.2 4i 22% >100 6b 79% 3.4 | 4g | 52% | >100 | 5g | 48% | 38.9 |
| | 4ĥ | 44% | 82.0 | 6a | 59% | 13.2 |
| | 4i | 22% | >100 | 6b | 79% | 3.4 |
| 4 j 27% 17.1 | 4j | 27% | 17.1 | | | |

CONCLUSION

We have accomplished the synthesis and the larvicidal evaluation tests for nineteen new isoxazole derivatives. These substances were even more potent against *A. aegypti* larvae than the ones earlier synthesized by our research group.⁴

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