





Sequential organocatalytic β-hydroxy and Mukayama addition: A perspective for the synthesis of Orlistat

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Keywords: Total synthesis, organocatalysis,Orlistat

INTRODUCTION

Orlistat is a reduced form of the natural product lipstatin, which is an antiobesity agent marked under the tradename of Xenical approved by the FDA as the first over-the-counter weight-loss medication.



Figure 1: Retrosynthetic analysis

RESULTS AND DISCUSSION

Firstly, the addition was mediated by 10 mol% of catalyst and the reaction condition (temperature and solvent) were optimized. The results are summarized in Table 1.

Table 1: Optimization of the reaction of conjugate addition of oxime to α - β -insaturated aldehyde.



 a Performed with 1 (0,05 mmol) 2 (0,5mmol), 3 (1,5 mmol), 4 (0,05mmol) in 0,25 mL of solvent. bDetermined by chiral stationary phase HPLC.

When the reaction was performed in THF as solvent, the $\beta\text{-hydroxy}$ carbonyl compound could be obtained

with 90% ee and 56% yield (Entry 3). Decreasing the reaction temperature to -10 °C and -20 °C (entries 5 and 6) furnished desired product in 96% and 91% ee respectively. The Mukayama aldol reaction with silyl enol ether were carried out using 50 mol % of TBAF, figure 2.

Figue 2: Screening of differents silyl enol ether



diastereoisomeric ratio of 75:25 and 89:11. Further optimization are under evaluation to improve the diastereoselectivity as well as the application or mercapto pyridinil enol ether for the synthesis of β -lactone ring.



In conclusion, the have optimized the application of organocatalysts **1** as a perfect mediator in the enantioselective addition of oxime α - β -unsaturated aldehyde. The nature of the solvent is very important once the one-pot sequential Mukayama additions are under evaluation.

ACKNOWLEDGEMENTS

The authors thank FAPESP (09/07281-0) and CNPq

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14th Brazilian Meeting on Organic Synthesis – 14th BMOS – September 01-05, 2011-Brasilia, Brazil