





Synthesis of Triazoles with Potential Antileishmanial Activity by Click Chemistry

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Keywords: Click Chemistry, Triazoles, Organic Synthesis.

INTRODUCTION

Click Chemistry is an organic synthesis procedure, which follows the principles of green chemistry, being introduced by Sharpless in 2001.¹ The most prominent chemical reaction using this procedure is the Cu(I)-catalyzed 1,3-dipolar azide-alkyne cycloaddition (CuAAC), leading to 1,2,3 triazoles.² From this context, our approach aims to synthesize substances with potential antileishmanial activity similar as presented by Silva,³ using Click Chemistry but with innovation as the insertion of the triazole unit is presented below:



Scheme 1. General structure of bioactive triazoles.

RESULTS AND DISCUSSION

Primarily, we did a test reaction in order to ascertain the efficiency of the Click reaction with similar procedure described by Nierengarten⁴ in 2008. The alcohol-bistriazole (II) was produced with 28% yield as a white solid.



Scheme 2. Synthesis of the alcohol-bistriazole II.

Our strategy to synthesize the alcohol-triazole is shown below:





Scheme 3. Synthetic route to hydroxyl-triazoles.

The synthesis began by chemoselective reduction of dimethyl 3-oxo-glutarate 1 using NaBH₄ and citric acid to afford dimethyl 3-hydroxy-glutarate 2. The sequence of preparation required protection of secondary alcohol and reduction of esters moieties, performed with DHP on acid catalysis (PTSA) followed by treatment of crude with LiAlH₄ in THF, respectively. The symmetrical diol 3, obtained in 50% for two steps, was monoprotected with TBSCI, leading to the alcohol 4. The next stage was the hydroxyl protection with TsCl in the presence of DMAP and Et₃N in DCM, leading to the formation of fully protected **5**. Azido treatment in hot acetonitrile yielded 6. The cycloaddition of azide compound 6 with hept-1-yno in heterogeneous mixture of DCM and water with use of CuSO₄ as catalyst produced the alcohol-triazole 7 in 33%.

CONCLUSION

We synthesized the alcohol-triazole **7** from dimethyl 3-oxo-glutarate in 8 steps, using Click Chemistry procedure as key step in cycloaddition between azide **6** and hept-1-yne.

ACKNOWLEDGEMENTS

GPSQ for chemical support and Capes for the scholarship.

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14th Brazilian Meeting on Organic Synthesis – 14th BMOS – September 01-05, 2011-Brasilia, Brazil