

Phenanthridine derivatives via palladium catalyzed intramolecular functionalization of C(sp²)-H bonds.

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INTRODUCTION

The biaryl structural motif is found in many natural and synthetic products of which phenanthridine derivatives are a specific heterocyclic system.¹

Biaryl bond formation is an important aspect in the synthesis of these compounds. As a result, organic chemists have sought to develop new and more efficient biaryl bond forming methods such as C-H direct arylation.²

RESULTS AND DISCUSSION

In developing upon our original methodology (M1),³ and that of others (M2),⁴ we have discovered a more robust method (M3) for the intramolecular C-H functionalization of *N*-alkyl- and *N*-aroyl- anilines to give phenanthridine derivatives (Figures 1 and 2).

Methodologies tested for biaryl bond formation.

All methods: Substrate (1 mmol), DMF (5 mL) and stirring at 120°C. M1 - 10 mol% $Pd(OAc)_2$, Bu_4NBr (1.1 mmol), KOAc (5 mmol). M2 - 5 mol% $Pd(OAc)_2$, 5 mol% dppe , K_2CO_3 (5 mmol) an argon atmosphere. M3 - 5 mol% $Pd(OAc)_2$, 5 mol% dppe, 5 mol% hydroquinone (HQ), Bu_4NBr (1.1 mmol), KOAc (5 mmol), an argon atmosphere.

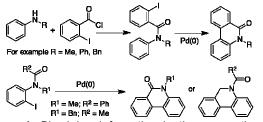


Figure 1. Biaryl bond formation in the preparation of phenanthridine derivatives via C-H functionalization.

Substrates for investigation of the C-H functionalization reaction were prepared by the acylation of appropriately functionlized aniline derivatives. Reaction methods **M2** and **M3** (absence of HQ) were found to have long induction periods when virgin glass vials and stirrer bars were used and gave principally products resulting from reduction of the C-I bond. However, the addition of HQ in **M3** resulted in "instaneous" reactions to give

the desired products. HQ had no effect upon **M2**. However, **M2** gave results comprable to **M3** when using glassware previously used for palladium catalyzed reactions. All products were spectroscopically characterized.

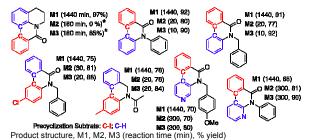


Figure 2. Illustrative results for catalytic intramolecular C-H functionalization. * The reaction times for M2 and M3 were determined by GC-MS analysis of reaction aliquotes.

* In the case of **M2** no cyclized product was formed and substrate was consumed due to reduction of the C-I bond.

CONCLUSION

The palladium catalyzed functionalization of a C-H bond allowed the satisfactory synthesis of phenanthridine derivatives. **M3** was found to be as good as or better than **M2** and both methods are a phenomenal advance upon our original method **M1**. Reaction times have been reduced from 12-24 hours to 10-300 minutes and catalyst loading has been halfed, the overall result being similar or better product yields.

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