



# Studies on the enantioselective Heck arylation of *N*-carboethoxy 3-pyrroline with aryldiazonium salts

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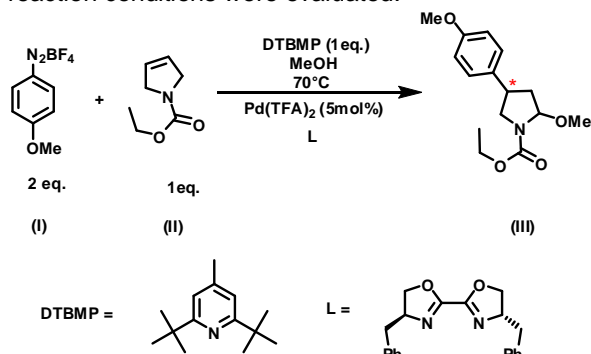
**Keywords:** Heck-Matsuda, enantioselective catalysis, *N*-carboethoxy 3-pyrroline

## INTRODUCTION

The Heck-Matsuda reaction is an useful methodology to create new C-C bonds from olefins and aryldiazonium salts as substrates.<sup>1</sup> There are very few reports describing this enantioselective reaction.<sup>2,3</sup> Because of the great interest in the synthesis of chiral compounds, new enantioselective methods are welcome. Since the first example of the enantioselective Heck-Matsuda reaction was reported,<sup>2</sup> the Correia's group has been intensely studying this reaction. In this context, arylation of substrates such as *N*-carboethoxy 3-pyrroline (**II**) was elected as a key challenge. Once synthesized, these Heck adducts could be used as a building block to access enantioenriched biologically active compounds.<sup>4</sup>

## RESULTS AND DISCUSSION

Previously established conditions<sup>2</sup> were used to prepare the Heck adduct (**III**) (Scheme 1). With this protocol the Heck product was obtained in only 6% yield with an ee of 51%. In view of the low yield, new reaction conditions were evaluated.



**Scheme 1.** Initial Heck-Matsuda reaction conditions

This reaction was also performed with different heating sources (microwave and sealed reactor), olefin concentration (2-4 equivalents with respect to the aryldiazonium salt) and reaction times (10-60 min). Improvements in yields were observed by increasing reaction time, but ee's were not affected significantly. Changes in olefin concentration do not affect the yield neither the ee's in a range of 2 – 3.5 equivalents. Application of microwave heating or performing the reaction in a sealed reactor for 1h,

allowed formation of the desired product in 35% yield and 40% ee.

Different solvents were evaluated and the best condition was obtained using MeOH (entry 1). Another promising condition is the system H<sub>2</sub>O:MeOH (entry 8). This system gave an excellent yield, although the ee was very poor.

**Table 1.** reaction scope in different solvents

Entry	Solvent	Yield (%)	e.e (%)
1	MeOH	35	40
2	Diox:MeOH (10%)	4	—
3	THF:MeOH (10%)	—	—
4	PhMe:MeOH (10%)	—	—
5	DMSO:MeOH (10%)	67	7
6	DMF:MeOH (10%)	54	12
7	H <sub>2</sub> O:MeOH (10%)	99	9

## CONCLUSION

In this work, the product of interest (**III**) was obtained in an excellent yield, but with rather low ee (Table, entry 8), or in moderated yield and ee (entry 1). Others reaction parameters are under evaluation, such as the palladium source and the base aiming at improving the enantioselectivity of the Heck-Matsuda reaction.

## ACKNOWLEDGEMENTS

We are grateful to FAPESP, CNPq, and CAPES.

## REFERENCES

- Kikukawa, K.; Matsuda, T. *Chem. Lett.* **1977**, 159.
- Correia, C. R. D.; Oliveira, C. C.; Salles Jr., A. G.; Santos, E. A. F., *Tetrahedron Lett.* **2012**, 53, 3325-3328.
- Werner, E.W.; Mei, T.S.; Burckle, A.J.; Sigman, M.S. *Science*, **2012**, 388, 1455-1458.
- Carpes, M. J. S.; Correia, C. R. D. *Tetrahedron Lett.* **2002**, 43, 741-744.