

1,3-Dipolar cycloadditions for the synthesis of low aggregation chlorins: A methodological study

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INTRODUCTION

Porphyrins are a class of heterocyclic compounds with comprehensive applications in technology, catalysis, materials science as well as in medicine.1 Chlorin derivatives are natural or synthetic porphyrinoids with one of the pyrrole rings reduced at the β position. So far no synthetic approach has proven to be as efficient in producing chlorins as cycloaddition reactions. For example, five member rings formed by the 1,3-dipolar cycloadditions on porphyrins,² are essential to produce compounds with low aggregation in solution, thus providing new molecules for use in photodynamic therapy (PDT) studies. Therefore, this work aims at the synthesis of new chlorin derivatives with low aggregation. Several conditions are being tested in order to provide a comprehensive study on the reactivity of dipoles (azomethine ylide) and dipolarophiles (porphyrins) under mild conditions.

RESULTS AND DISCUSSION

Initially, we synthesized the amine 4 in two wellknown steps (Scheme 1).3





Synthesis meso-tetrakis(pentafluorophenyl) of porphyrin (TPPF) (5) was performed using pentafluorobenzaldehyde and pyrrole, obtaining TPPF in 8% yield after laborious purifications. Metallation of 5 with NiCl₂.6H₂O in DMSO furnished 6 in 95% yield.4

After preparation of the metallated porphyrin 6, we studied the 1,3-dipolar reaction with one-pot generated azomethine ylide 7 at room temperature, under several dipole concentrations, obtaining the chlorin 8 and the isobacteriochlorin 9 in different proportions (Table 1).



Scheme 2. Products of the cycloaddition of porphyrin 6. Table 1. Yields obtained with different amounts of 4.

Equivalents of 4 (dipole precursor)	Proc	Product Yields (%)		
	6	8	9	
5	80.3	10	0	
10	63	22.4	traces	
15	53	36.6	5.3	
20	46	40	6.1	
25	36.7	45.8	12.4	
30	0	traces	65.5	

CONCLUSION

Chlorin 8 and isobacteriochlorin 9 derivatives have been obtained under mild conditions using a 1,3dipolar cycloaddition procedure. Also, compounds 8 and 9 revealed low aggregation in solution due to the "L" shape of these molecules.^{2c} Currently, we are finishing up the characterizations, and evaluating the photophysical properties of such products.

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