



Synthesis and antioxidant activity of nitroaryl-1,2,3-triazoles

Wagner O. Valença,* Pollyanna L. F. da Costa, Jucleiton J. R. de Freitas, Tânia Maria S. da Silva, Ronaldo N. de Oliveira *

Departamento de Ciências Moleculares, Universidade Federal Rural de Pernambuco, Recife, PE, Brazil

*E-mails: wagnerufrpe@yahoo.com.br; ronaldonoliveira@dcm.ufrpe.br

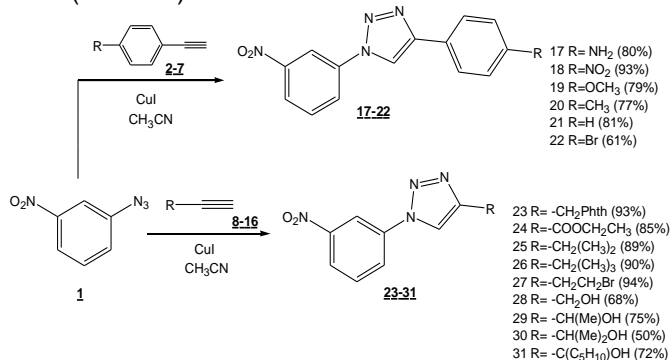
Keywords: 1,2,3-triazole, cycloaddition, antioxidant activity

INTRODUCTION

1,2,3-Triazole derivatives are known to exhibit a range of biological activities such as, anti-inflammatory, antileishmanial and anti-*T. cruzi*.¹⁻³ The nitroaryls and nitroheterocycles are considered to be medicinally a valuable group of compounds, broadly redox properties of nitroaromatic compounds are associated to biological activities.⁴ In our antioxidative screening study of nitroaryl-1,2,3-triazoles using free radical scavenging activity of DPPH and ABTS, we report here some active compounds.

RESULTS AND DISCUSSION

We performed the synthesis of bis-aryl-1,2,3-triazoles **17-22** and a variety of functional groups **23-31** linked via 1,3-dipolar cycloaddition reaction (1,3-DCR). The reaction between 1 mmol of nitroaryl-azide **1** and 1.5 mmol of terminal alkynes **2-16** was carried out in the presence of 10 mol% CuI, using CH₃CN as solvent and stirring at room temperature under argon atmosphere for 24 h.⁵ The compounds were obtained in moderate to good yields of 50-94% (Scheme).



Scheme. Synthesis of the nitroaryl-1H-1,2,3-triazole **17-31**

All the compounds have been tested their scavenging activity (%SA) and their antioxidant concentration required to inhibit 50% of the radicals (EC₅₀).⁶ Only the compounds containing functional groups that were susceptible to oxidation, such as hydroxyl **28-29** and amine **17**, showed the best antioxidant activities (Table). These results may be rationalized because the 1,2,3-triazole group is

known as stable, and now we have described their stability under scavenging conditions.

Table. The best results for antioxidant activities of nitroaryl-1H-1,2,3-triazoles.

Compounds	EC ₅₀ (μg/mL) ^a	
	DPPH ^b	ABTS ^b
17	7.79 ± 1.68	13.42 ± 0.97
28	49.75 ± 1.37	32.98 ± 0.39
29	47.97 ± 5.42	22.23 ± 0.07
Ascorbic acid	1.67 ± 0.02	-
TROLOX	-	3.86 ± 0.04

^a Antioxidant concentration required to reduce the original radical population by 50%. ^b Values represent mean ± standard deviation: n = 3.

CONCLUSION

We have synthesized a series of nitroaryl-1H-1,2,3-triazoles **17-31** in moderate to good yields. Only the compounds **17**, **28** and **29** presented satisfactory results as an antioxidant. These three compounds have been found to be a lead antioxidant agent for further study.

ACKNOWLEDGEMENTS

The authors are grateful to CNPq, FACEPE-PRONEM for financial support and CAPES for providing a fellowship to one of us (W.O.V.). Our thanks are also due to Analytical Centers CENAPESQ-UFRPE and DQF-UFPE.

REFERENCES

- Kumar, S.S.; Kavitha, H. P.; *Mini-Rev. Org. Chem.* **2013**, *10*, 40;
- Agalave, S. G.; Maujan, S. R.; Pore, V. S.; *Chem. Asian J.* **2011**, *6*, 2696.
- Assis, S. P. O.; da Silva, M. T.; de Oliveira, R. N.; Lima, V. L. M.; *The Scientific World Journal* **2012**, ID 925925, DOI: 10.1100/2012/925925.
- Guimarães, T. T.; Pinto, M. C. F. R.; Lanza, J. S.; Melo, M. N.; Monte-Neto, R. L.; de Melo, I. M. M.; Diogo, E. B. T.; Ferreira, V. F.; Camara, C. A.; Valença, W. O.; de Oliveira, R. N.; Frézard, F.; da Silva Jr, E. N.; *Eur. J. Med. Chem.* **2013**, *63*, 523.
- Wardman, P.; *Environ. Health Perspect.* **1985**, *64*, 309.
- Valença, W. O.; de Freitas, J. J. R.; de Oliveira, R. N.; *Orbital Elec. J. Chem.* **2012**, *4* (Suppl. 1), 31.
- Souza, S.A.; Camara, C. A.; da Silva, E. M. S.; da Silva, T. M. S.; *Evidence-Based Complementary and Alternative Medicine* **2013**, ID 801383, DOI: http://dx.doi.org/10.1155/2013/801383.