

SYNTHESIS AND CHARACTERIZATION OF NANOSCALE HYDROXYAPATITE

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Synthetic nanohydroxyapatite (nHA) is a prominent material to be applied in bone tissue engineering devices, due to nHA similarity with the main component of the bone inorganic phase, biocompatibility, biodegradability, and bioactivity. The specific characteristics of the nHA crystals are dependent on the synthesis method. Therefore, the aim of this work was to evaluate the structure of the nHA obtained through chemical precipitation, for the understanding of nHA physico-chemical and biological properties. nHA was produced by dropwise addition of 1600 mL of 0.7 M aqueous Ca(NO₃)₂.4H₂O (pH 5.5) on 1100 mL of 0.5 M (NH₄)₂HPO₄ (pH 10.4, adjusted with concentrated NH₃) under stirring at 80°C. The reaction mixture was aged 24 h then vacuum filtered and washed with water and ethanol. The precipitated nHA was dried at 80 °C for 24 h. The structure obtained nHA was characterized by X-Ray Diffraction and the Rietveld refinement method, Transmission Electron Microscopy and Fourier Transform Infrared Spectroscopy. Preliminary results indicate that the material consisted of calcium deficient hydroxyapatite nanocrystals with lattice parameters a=b=9.43019 Å and c=6.88162 Å, and c-axis preferentially oriented. Nanoparticles, shaped as rods, presented mean crystallite size of ~21 nm (~47 nm length and ~8 nm width) and specific surface area of 90.1 m²/g. According to the obtained results the method of synthesis of nHA seems to be reproducible and effective to prepare large quantities of nHA to be evaluated as biomaterials.